Docket No.: 043888-0267

PATENT

R:290

In re Application of : Customer Number: 20277

Tsutomu OHZUKU, et al. : Confirmation Number: 9492

Application No.: 10/629,815 : Group Art Unit: 1745

Filed: July 30, 2003 : Examiner: LEE, CYNTHIA K

For: POSITIVE ELECTRODE ACTIVE MATERIAL AND NON-AQUEOUS ELECTROLYTE

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

SECONDARY BATTERY CONTAINING THE SAME

Declaration Under 37 C.F.R. § 1.132

Mail Stop Amendment Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

- I, Dr. Hiroshi Yoshizawa, declare as follows:
- I received a doctorate of Engineering from the Graduate School of Engineering,
 Osaka City University.
 - 2. My field of specialty is lithium ion secondary batteries.
- 3. Since 1987, I have been employed by Matsushita Battery Industrial Co., Ltd., a subsidiary of Matsushita Electric Industrial Co., Ltd.
- 4. I have been working in the field of research and development of lithium ion secondary batteries for the past 10 years.

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- 5. I am a coinventor of U.S. Patent Application Serial No. 10/629,815, POSITIVE ELECTRODE ACTIVE MATERIAL AND NON-AQUEOUS ELECTROLYTE SECONDARY BATTERY CONTAINING THE SAME, filed July 30, 2003 (the present invention).
- 6. I have read and am familiar with the disclosure of the above-captioned patent application. I have read and am familiar with Ohzuku et al., Layered Lithium Insertion Material of LiCo_{1/3}Ni_{1/3}Mn_{1/3}O₂ for Lithium-Ion Batteries (Chemistry Letters, CL-010390, Vol. 30 (2001), No. 7, pp. 642-43) and Ohzuku et al., Layered Lithium Insertion Material of LiN_{1/2}Mn_{1/2}O₂ for Lithium-Ion Batteries: a Possible Alternative to LiCoO₂ for Advanced Lithium-Ion Batteries (Chemistry Letters, CL-010410, Vol. 30 (2001), No. 8, pp. 744-45).
- 7. At my direction and under my supervision positive electrode active material according to Example 1-2 of the present invention was fabricated.
- 8. At my direction and under my supervision positive electrode active material according to Layered Lithium Insertion Material of LiCo_{1/3}Ni_{1/3}Mn_{1/3}O₂ for Lithium-Ion Batteries (Chemistry Letters, CL-010390) was fabricated from LiOH·H₂O, CoCO₃, and nickel manganese hydroxide in the manner described on page 642, left column, 17th line from the bottom, et seq.
- 9. The positive electrode active material fabricated according to the present invention and the positive electrode active material fabricated according to Chemistry Letter, CL-010390 were analyzed by Electron Probe Micro Analysis (EPMA) using a JEOL Ltd. JXA-8900, under the following conditions:

Temperature:

Room Temperature

Irradiation Current:

 $5 \times 10^{-8} \, \text{A}$

Acceleration Voltage:

15 kV

Range of Analysis:

(x-direction) $0.2 \mu m \times 340 \text{ points} = 68 \mu m$

(y-direction) 0.2 μ m × 255 points = 51 μ m.

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10. The distribution of cobalt in the Chemistry Letters, CL-010390 material is shown in Exhibit A in the micrograph labeled "prior art" and the cobalt distribution in the material according to the present invention is shown in the micrograph labeled "our invention." Green represents a low concentration and red represents a high concentration.

- 11. Exhibit B illustrates the distribution of cobalt, manganese, and nickel in the material according to the present invention.
- As illustrated in Exhibit A, positive electrode active material fabricated according 12. to Chemistry Letters, CL-010390 exhibit a non-uniform elemental distribution. Exhibit A, as indicated by the Co-rich and Co-poor areas, clearly shows a widely varying distribution of cobalt in the Chemistry Letters, CL-010390 positive electrode active material. In contrast thereto. Exhibits A and B clearly illustrate that cobalt is uniformly dispersed throughout the positive electrode active material according to the present invention.
- 13. I further declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true, and further, that these statements are made with the knowledge that willful false statements and the like are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of this application or any patent issuing therefrom.

January 18, 2007 Joshi Joshizawa
Hiroshi Yoshizawa